

## EUDESMANOLIDES FROM *CALEA TRICHOMATA*

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**Key Word Index**—*Calea trichomata*; Asteraceae; Heliantheae; sesquiterpene lactones; eudesmanolides.

**Abstract**—Chemical analysis of *Calea trichomata* yielded, besides four known eudesmanolides, a new eudesmanolide, trichomatolide A. The structure of the new compound was established by chemical and spectral methods.

### INTRODUCTION

In continuation of our biochemical systematic study of members of the subtribe Galinsoginae, tribe Heliantheae [1] we have analysed *Calea trichomata* of section *Calea* from Chiapas, Mexico for their sesquiterpene lactone constituents. Besides the known 1 $\beta$ -hydroxy-8 $\beta$ -tigloxy-eudesman-6 $\alpha$ ,12-olide derivatives (1-4) which had previously been isolated from *Calea rotundifolia* [2] and *Liatris laevigata* [3], a new eudesmanolide was found. The

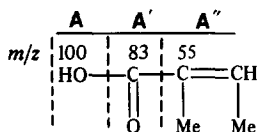
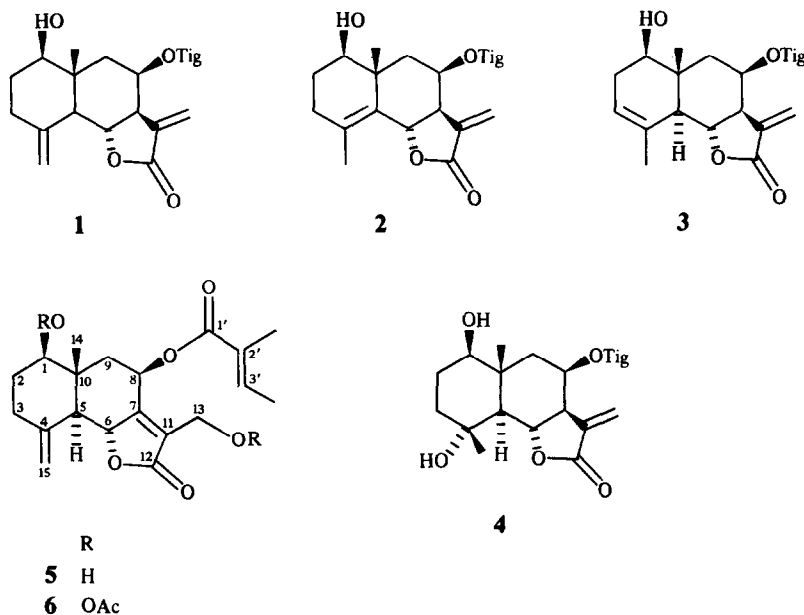
identities of the known lactones (1-4) were established by  $^1\text{H}$  NMR and mass spectral correlations [2, 3] and the new compound was characterized by chemical and spectral methods.

### RESULTS AND DISCUSSION

Trichomatolide A (5),  $\text{C}_{20}\text{H}_{26}\text{O}_6$ , is a gum with an IR spectrum showing the presence of hydroxyl groups (absorption bands at 3415 and 3325  $\text{cm}^{-1}$ ), a  $\gamma$ -lactone moiety (1745  $\text{cm}^{-1}$ ) and an  $\alpha,\beta$ -unsaturated ester(s) (1695  $\text{cm}^{-1}$ ). The ester function was assigned to a tiglate group on the basis of diagnostic  $^1\text{H}$  NMR signals (a one-proton quartet of quartets at  $\delta$ 6.89, and two three-proton methyl singlets at 1.82 and 1.81, respectively), together

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with the characteristic mass spectral peaks for tiglate side-chains at  $m/z$  83 (A') and 55 (A"). On the basis of the chemical shift of H-8 ( $\delta$ 6.19) the tiglate ester must be attached at C-8. Further assignments of the proton signals were deduced from detailed double irradiation experiments (Table 1).

The  $^1\text{H}$  NMR spectrum of compound **5** was very similar to the one of 8 $\beta$ -tigloxyreynosin (**1**) and in agreement with an eudesmanolide type skeleton, except for the following differences: (a) the typical signals for the exocyclic methylene lactone protons near  $\delta$ 6 were missing, and instead, a two-proton doublet at 4.57 ( $J = 1.5$  Hz) was found; (b) the H-6 signal did not appear as a triplet as in compounds **1**–**4**, but as a broad doublet at 5.13 ( $J = 11.0$  Hz); (c) there was no H-7 signal in the spectrum of **5**.

These differences between the  $^1\text{H}$  NMR spectrum of compound **5** and the spectra of the other four eudesmanolides (**1**–**4**) suggested an arbusculin D type lactone ring [4] with an endocyclic 7(11)-double bond and a saturated C-13. Moreover, the chemical shift of the two-proton signal H-13 ( $\delta$ 4.57) suggested the presence of a hydroxyl group at C-13. Acetylation of compound **5** provided the diacetate **6**,  $\text{C}_{24}\text{H}_{32}\text{O}_8$ , which lacked hydroxyl absorptions in the IR spectrum, but instead showed an additional carbonyl band at  $1735\text{ cm}^{-1}$  which was assigned to the acetate groups. The  $^1\text{H}$  NMR spectrum of the diacetate **6** (Table 1) showed a paramagnetic shift of H-1 from  $\delta$ 3.44 to 4.65, and a shift of the two-proton H-13 doublet which now appeared as two doublets centered at 5.19 ( $J = 14.0$  Hz) and 4.74 ( $J = 14.0$ ; 1.8 Hz) in a typical AB pattern. These paramagnetic shifts in the diacetate **6** confirmed the presence of hydroxyl groups on C-1 and C-13 in compound **5**.

The proposed structure for lactone **5** was in agreement with the obtained  $^{13}\text{C}$  NMR data given in the Experimental. Comparison of these data with the ones reported for 8 $\beta$ -tigloxyreynosin (**1**) [3] showed two major dif-

ferences: (a) the C-7 signal, which was a doublet at  $\delta$ 53.39 in **1** [3] appeared as a singlet at 157.1 in **5** confirming the quarternary and olefinic character of C-7 in **5**. (b) C-13 which was a triplet at  $\delta$ 119.49 in **1** underwent a diamagnetic shift to 56.9 in compound **5** indicating that C-13 represents a saturated carbon atom bearing an hydroxyl group.

The stereochemistry at C-1 and C-8 was assigned by correlation of the dihedral angles of the protons with the experimentally measured coupling constants. The large coupling ( $J_{1,2\beta} = 11.2$  Hz) suggested an equatorial,  $\beta$ -oriented hydroxyl group at C-1; similarly, the small coupling constants  $J_{8,9a} = 2.0$  Hz and  $J_{8,9b} = 4.6$  Hz indicated that the tiglate ester substituent at C-8 be in a  $\beta$  position. The large splitting ( $J_{5,6} = 11$  Hz) was in accord with an antiperiplanar arrangement of the protons at C-5 and C-6.

## EXPERIMENTAL

*Calea trichomata* D. Smith was collected on July 29, 1978 in Chiapas, Mexico, 2.8 miles south of La Trinitaria along Highway 190 (L. Urbatsch, No. 3335, voucher deposited at LSU, U.S.A.). The air-dried plant material (652 g) was extracted and worked up as previously described [5], providing 9.1 g of crude syrup which was chromatographed on a silica gel column with petrol– $\text{Me}_2\text{CO}$  mixtures of increasing polarity; 38 fractions of 200 ml each were taken.

Fractions 16–18 (1.4 g) were re-chromatographed on a silica gel column with petrol–EtOAc mixtures of increasing polarity; 25 fractions of 100 ml each were taken. Fractions 11–14 afforded 230 mg of **1**, 20 mg of **2**, and 7 mg of **3**. Fractions 15–16 provided 65 mg **5**, and fraction 17–18 gave 78 mg **4**. The  $^1\text{H}$  NMR parameters of the known lactones **1**–**4** were identical with the ones of the compounds described in the literature [2, 3].

*Trichomatolide A* (**5**).  $\text{C}_{20}\text{H}_{26}\text{O}_6$ , gum; UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 220 ( $\epsilon 1.14 \times 10^4$ ); CD (MeOH;  $c 4.56 \times 10^{-3}$ ):  $[\theta]_{280}^{25} + 2.9 \times 10^2$ ,  $[\theta]_{243}^{25} - 8.02 \times 10^4$ ; IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3415 (OH), 3325 (OH),

Table 1.  $^1\text{H}$  NMR spectral data of compounds **4** and **6** (200 MHz,  $\text{CDCl}_3$ , TMS as int. standard)

|         | 5   | 6   |
|---------|---|---|
| H-1     | 3.44 <i>dd</i> ( $J = 11.2$ ; 5.0 Hz)       | 4.65 <i>dd</i> ( $J = 12.0$ ; 5.0 Hz)       |
| H-2a    | 1.66 <i>m</i>                               | 1.55–1.95*                                  |
| H-2b    |   |   |
| H-3a    | 2.38 <i>ddd</i> ( $J = 14.0$ ; 5.0; 2.0 Hz) | 2.40 <i>ddd</i> ( $J = 15.0$ ; 5.6; 2.2 Hz) |
| H-3b    | 2.0–2.1*                                    | —   |
| H-5     | 1.81 <i>d</i> ( $J = 11.0$ Hz)              | 1.91 <i>d</i> ( $J = 11.0$ Hz)              |
| H-6     | 5.13 <i>ddd</i> ( $J = 11.0$ ; 1.0; 1.0 Hz) | 5.17 <i>d</i> ( $J = 11.0$ Hz)              |
| H-8     | 6.19 <i>dd</i> ( $J = 4.6$ ; 2.0 Hz)        | 6.17 <i>dd</i> ( $J = 4.8$ ; 2.0 Hz)        |
| H-9a    | 2.58 <i>dd</i> ( $J = 15.5$ ; 2.0 Hz)       | 2.27 <i>dd</i> ( $J = 16.0$ ; 2.0 Hz)       |
| H-9b    | 1.66 <i>dd</i> ( $J = 15.5$ ; 4.6 Hz)       | 1.64 <i>dd</i> ( $J = 16.0$ ; 4.8 Hz)       |
| H-13a   | 4.57 <i>d</i> ( $J = 1.5$ Hz)               | 5.19 <i>d</i> ( $J = 14.0$ Hz)              |
| H-13b   |   | 4.74 <i>dd</i> ( $J = 14.0$ ; 1.8 Hz)       |
| C-10-Me | 1.13 <i>s</i>                               | 1.19 <i>s</i>                               |
| H-15a   | 5.06 <i>br s</i>                            | 5.09 <i>br s</i>                            |
| H-15b   | 5.00 <i>br s</i>                            | 5.04 <i>br s</i>                            |
| H-3'    | 6.89 <i>qq</i> ( $J = 7.2$ ; 1.8 Hz)        | 6.84 <i>qq</i> ( $J = 7.8$ ; 2.0 Hz)        |
| C-3'-Me | 1.82 <i>br</i>                              | 1.83 <i>br</i>                              |
| C-2'-Me | 1.81 <i>s</i>                               | 1.80 <i>s</i>                               |

\*Obscured by other signals.

1745 ( $\gamma$ -lactone), 1695 (conj. ester), 1650 (double bond); EIMS (probe)  $m/z$  (rel. int.): 362  $[M]^+$  (2.6), 262  $[M-A]^+$  (41.1), 244  $[M-A-H_2O]^+$  (12.5), 200  $[M-A-CO_2]^+$  (10.6), 83  $[A']^+$  (100), 55  $[A'']^+$  (39.8);  $^{13}C$  NMR (50.32 MHz,  $CDCl_3$ , TMS as internal standard): 76.8  $d$  (C-1), 31.2  $t$  (C-2), 33.7  $t$  (C-3), 142.3 (C-4), 55.3  $d$  (C-5), 78.6  $d$  (C-6), 157.1  $s$  (C-7), 65.5  $d$  (C-8), 41.4  $t$  (C-9), 41.1  $s$  (C-10), 128.1  $s$  (C-11), 172.4  $s$  (C-12), 56.9  $t$  (C-13), 12.2  $q$  (C-14), 110.8  $t$  (C-15), 168.2  $s$  (C-1'), 128.2  $s$  (C-2'), 140.0  $d$  (C-3'), 14.8  $q$  (C-2'-Me), 13.0  $q$  (C-3'-Me). (Calc. for  $C_{20}H_{26}O_6$ : 362.1693. Found: MS 362.1675.)

*Trichomatolide A diacetate* (6). Acetylation of 10 mg 5 in pyridine- $Ac_2O$  for 20 hr, followed by usual work-up, gave the diacetate (6),  $C_{24}H_{32}O_8$ , gum; IR  $\nu_{max}^{CHCl_3}$   $cm^{-1}$ : 1770 ( $\gamma$ -lactone), 1735 (acetate, ester), 1720 (conj. ester), 1655 (double bond); EIMS (probe)  $m/z$  (rel. int.): 446  $[M]^+$  (4), 386  $[M-HOAc]^+$  (4), 304  $[M-A-C_2H_2O]^+$  (4.0), 286  $[M-A-HOAc]^+$  (4.3), 244  $[M-HOAc-A-C_2H_2O]^+$  (21.5), 226  $[M-2HOAc-A]^+$

(35.3), 211  $[M-HOAc-A-Me]^+$  (17.1), 83  $[A']^+$  (100), 55  $[A'']^+$  (28.4), 43  $[Ac]^+$  (43.4).

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#### REFERENCES

1. Stuessy, T. F. (1977) in *The Biology and Chemistry of the Compositae* (Heywood, V. H., Harborne, J. B. and Turner, B. L., eds). Academic Press, London.
2. Bohlmann, F., Gupta, R. K., Jakupovic, J., King, R. M. and Robinson, H. (1981) *Phytochemistry* **20**, 1635.
3. Herz, W. and Kulanthaivel, P. (1983) *Phytochemistry* **22**, 715.
4. Irwin, M. A. and Geissman, T. A. (1973) *Phytochemistry* **12**, 853.
5. Fischer, N. H., Wiley, R. A., Lin, H. N., Karimian, K. and Politz, S. M. (1975) *Phytochemistry* **14**, 2241.

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## INTEGRIFOLIN, A GUAIANOLIDE FROM *ANDRYALA INTEGRIFOLIA*\*

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**Key Word Index**—*Andryala integrifolia*; Compositae; Lactuceae; sesquiterpene lactones; guaianolide; integrifolin.

**Abstract**—Integrifolin, the major constituent of *Andryala integrifolia*, has been isolated and characterized as 3 $\beta$ ,8 $\beta$ -dihydroxy-4(15),10(14),11(13)-trien-(1 $\alpha$ H), (5 $\alpha$ H) guaian-6,12-olide (8-epi-desacylcynaropicrin).

#### INTRODUCTION

Only one species of the genus *Andryala* (tribe Lactuceae) has been investigated chemically [1, 2]. We have now initiated the study of the constituents of *A. integrifolia* L., a species found in mediterranean Europe [3]. The main constituent in this plant is a sesquiterpene lactone of the guaiane series, which has been named integrifolin (1a). In addition, the flavonoids luteolin [4] and apigenin [5] were isolated.

#### RESULTS AND DISCUSSION

Integrifolin, mp 206–208°,  $[\alpha]_D -17.5^\circ$ , IR  $\nu_{max}^{KBr}$   $cm^{-1}$ : 3440 (OH), 1750 ( $\alpha,\beta$ -unsaturated- $\gamma$ -lactone ring), 1655, 1630 (double bonds); MS  $m/z$ : 262.120  $[M]^+$ , was obtained from the medium polar fractions. Its  $^1H$  NMR data (Table 1) showed it was 3 $\beta$ ,8 $\beta$ -dihydroxy-4(15),

10(14),11(13)-trien-(1 $\alpha$ -H,5 $\alpha$ -H)-guaian-6,12-olide (1a). The most characteristic features of this spectrum are signals of the  $\alpha$ -methylene- $\gamma$ -lactone grouping, two exocyclic methylenes (C-14 and C-15), and the C-6 lactonic

Table 1.  $^1H$  NMR spectral data for integrifolin 1a (ppm from TMS)

|      |      |      |       |      |      |
|------|------|------|-------|------|------|
| H-1  | 2.66 | ddd  | H-9   | 2.38 | dd   |
| H-2  | 1.52 | ddd  | H-9'  | 2.11 | dd   |
| H-2' | 1.98 | ddd  | H-13  | 5.42 | d    |
| H-3  | 4.28 | dddd | H-13' | 6.14 | d    |
| H-5  | 2.53 | dd   | H-14  | 4.71 | br s |
| H-6  | 4.32 | dd   | H-14' | 4.83 | br s |
| H-7  | 2.78 | dddd | H-15  | 5.09 | br s |
| H-8  | 4.08 | ddd  | H-15' | 5.20 | br s |

$J$  (Hz): 1,2 = 7.5; 1,2' = 9.5; 1,5 = 9.5; 2,2' = 13; 2,3 = 9; 2',3 = 7.5; 3,15 = 1.5; 5,6 = 10; 6,7 = 9; 7,8 = 3; 7,13' = 3.5; 7,13 = 3; 8,9 = 8, 9' = 6 and 9,9' = 13.5.

\*Part 1 in the series "Structure and Chemistry of Secondary Metabolites from Compositae".